

Synthetic Emergence in *N*-Arylimidazoles: A Review

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ABSTRACT Immense therapeutic potential of *N*-arylimidazoles prompted various researchers worldwide to develop different synthetic routes to these heterocycles. The present review highlights and summarises various synthetic approaches, conventional as well as greener protocols involved in the synthesis of *N*-arylimidazoles in the past two decades. The reported methods involve two basic approaches toward the synthesis of *N*-arylimidazoles (i) synthesis of imidazole moiety using open chain substrates and (ii) *N*-arylation of imidazoles.

KEYWORDS Azoles, Imidazoles, *N*-Arylimidazoles, Copper catalysts.

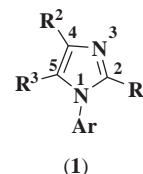
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INTRODUCTION

Imidazole belongs to azoles family and has been first synthesized by the reaction of glyoxal with ammonia. Initially, it was named as glyoxalin. It exists in two equivalent tautomeric forms at room temperature and possesses similar properties such as pyrrole and pyridine. It is an integral part of several natural products such as histamine, biotin, alkaloids, and nucleic acids. On fusion with pyrimidine, it forms purine ring which is the most widely occurring nitrogen-containing heterocycle in nature. Imidazole also provides functional group sites that could be modified to yield a variety of bioactive heterocycles.^[1-3] Imidazole and its derivatives have occupied unique place in the field of medicinal chemistry due to their diverse biological activities such as antibacterial,^[4] antifungal,^[5] anti-inflammatory,^[6] antimalarial,^[7] histamine-H₃ antagonist,^[8] farnesyl transferase and geranyl transferase-inhibitor,^[9] antioxidant,^[10] antitumor,^[11] antiparasitic,^[12] and antiprotozoal.^[13] Various drugs having substituted imidazole moiety are available in market such as dimetridazole,

azathioprine, carnidazole, mezlocillin, metronidazole, and mizoribine.

Among the imidazole derivatives, *N*-arylimidazoles (**1**) have shown immense potential as key structural motifs in medically important compounds, such as tyrosine kinase inhibitors,^[14,15] γ -secretase modulators,^[16] serotonin receptor antagonists,^[17] and glycine transporter type-1 inhibitors.^[18] These are also known to be an integral part of drugs that show adrenoreceptor α_2 -antagonist,^[19] histamine H₁ receptor antagonist,^[20-22] anxiolytic,^[23,24] antifungal properties, etc.^[25-28]



N-Arylimidazoles

Merely a single review is there on the bioactive derivatives of *N*-arylimidazoles,^[28] and to the best of our knowledge,

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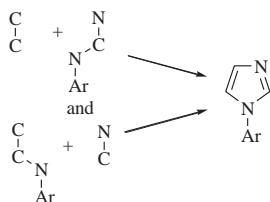
no review is yet reported on the synthesis and properties of *N*-arylimidazoles. Thus, it is worthwhile to prepare an updated detailed review on the same. The present review summarises the developments in the synthesis of *N*-arylimidazoles. The reported methods involve two basic approaches for the synthesis of *N*-arylimidazoles: (i) Constructing imidazole moiety from open chain substrates and (ii) *N*-arylation of imidazoles (with no substituent at position 1) **Chart 1**. Accordingly, the methods are described in two sections **2** and **3** starting from open chain compounds and imidazoles, respectively.

SYNTHESIS OF *N*-ARYLIMIDAZOLES FROM OPEN CHAIN SUBSTRATES

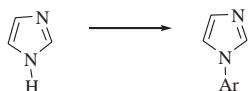
N-Arylimidazoles (**4**) have been synthesised in very high yields by introducing different aryl groups from the corresponding *N*-arylformamides (**3**) and *N*-formylglycine esters (**2**) at room temperature^[29] in the presence of proline ligand [**Scheme 1**].

N-Arylimidazoles (**7**) containing electron withdrawing group have been obtained in very good yields when aryl isocyanides (**5**) are treated with isocyanides (**6**) in the presence of 1,10-phenanthroline as a ligand^[30] in a copper-catalysed cross-cycloaddition [**Scheme 2**].

i) Construction of imidazole moiety from open chain substrates

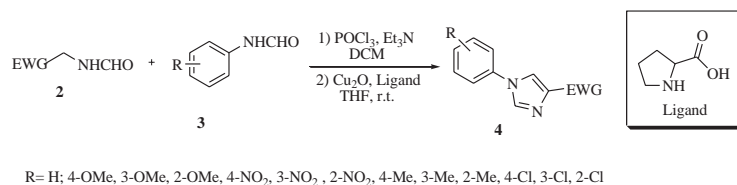


ii) *N*-Arylation of imidazoles

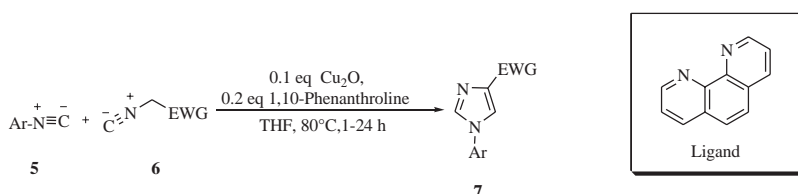


and more

Chart 1: Basic strategies for *N*-arylimidazoles



Scheme 1: *N*-Arylimidazoles from *N*-arylformamides and *N*-formylglycine esters



Scheme 2: *N*-Arylimidazoles from aryl isocyanides

Copper halides support a range of sensitive functional groups leading to high yielding synthesis of arylated imidazoles in the absence of specific ligands. The use of protic as well as aprotic solvent has led to successful *N*-arylation.

Pooi *et al.* reported a straightforward and high yielding synthesis of 1,4-diaryl-1*H*-imidazoles (**10**, **13**) from aryl isocyanides (**9**) and benzyl isocyanides (**8**) through NHC-Cu catalyzed isocyanide insertion into alcohol, forming *N*-arylformimidate (**12**) intermediate. The intermediate then subsequently promoted to the desired diarylimidazole (**13**) with **8** in the presence of base^[31] (KOtBu) [**Scheme 3**].

N-Arylimidazoles (**16**) have been obtained using arylacetic acids (**14**), *N*-arylbenzamidines (**15**), and nitroalkanes in the presence of 2,2'-bipyridyl as a ligand and DMF using copper sulfate as a catalyst^[32] [**Scheme 4**].

N-Arylimidazole quinoline-2-ones (**21**) were synthesized by condensing hydrazino carbonyl-phenyl vinylacetamide derivative (**17**) with coumarins in acetic acid by two different processes – a two-step process and another one-pot process^[33] [**Scheme 5**].

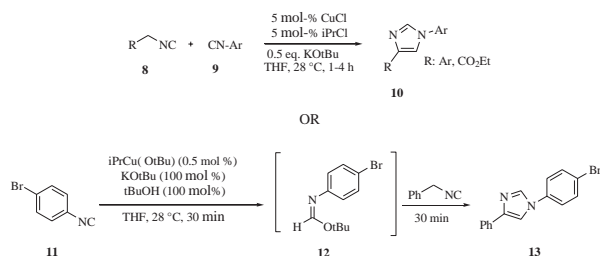
In another method, anilines (**22**) and ethyl glyoxylate (**23**) were treated in methanol to obtain α -anilino- α -methoxyacetates (**24**) which were cyclized with TosMIC (toluenesulfonylmethyl isocyanide) to give *N*-arylimidazole-5-carboxylates^[34] (**25**) [**Scheme 6**].

Imidazole-thiones (**28**) have been synthesized from aminoacetaldehyde diethyl acetal and arylisothiocyanate^[35] (**26**) [**Scheme 7**].

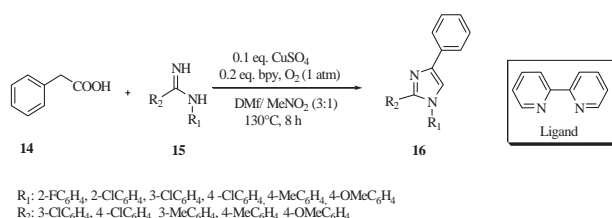
SYNTHESIS THROUGH *N*-ARYLATION OF IMIDAZOLES

Some traditional methods, which have been used for the synthesis of *N*-arylimidazoles, involve S_NAr substitution reaction of imidazoles with activated aryl halides or classical Ullmann-type coupling reaction.^[36,37] These methods suffer from serious drawbacks such as the use of stoichiometric amount of copper reagents, high reaction temperature, and long reaction time period. Therefore, further developments focused on the use of catalysts mainly Cu salts. Some *N*- and *O*-containing ligands have also been used to facilitate

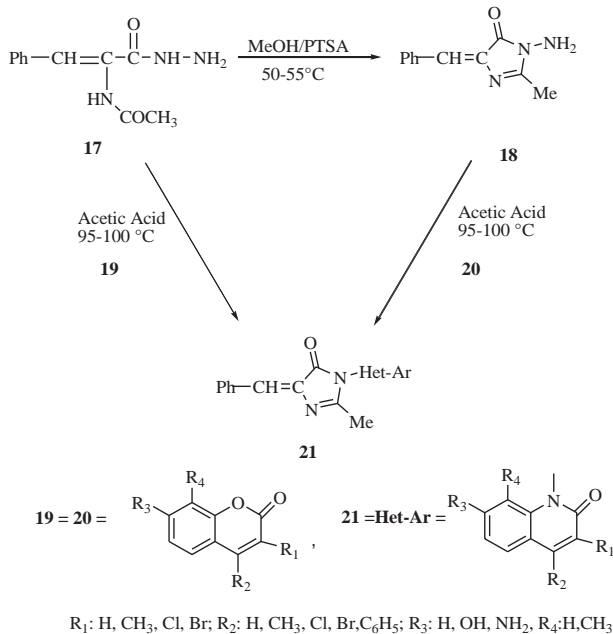
the copper catalysed C-N coupling reactions.^[38-42] Further, realising the potential of ligands, a series consisting of mono and bidentate ligands have been developed and tested to improve the efficiency of the reactions associated with Ullmann-type C-N bond formation.^[43-46] The Lam-Chan reaction (Cu-catalysed cross-coupling between imidazoles and arylboronic acids) has also been found a better alternative



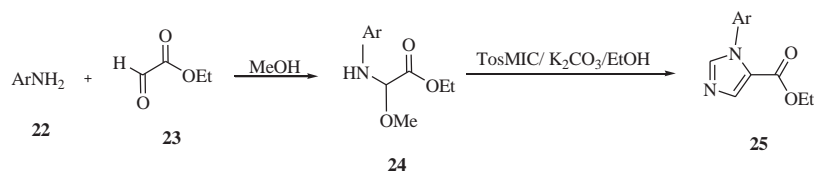
Scheme 3: *N*-arylimidazoles from benzyl isocyanide derivatives



Scheme 4: *N*-Arylimidazoles from *N*-arylbenzamidines



Scheme 5: *N*-arylimidazoles from coumarins



Scheme 6: *N*-Arylimidazoles from ethyl glyoxylate and aniline

because of low-temperature requirement^[47-51] but the selection of solvents,^[52] bases,^[53] additives,^[54] and substrate types^[55] often needs to be optimized. Therefore, the present section highlights the recent synthetic developments in *N*-arylimidazoles through catalysis. The synthetic methods have been classified and discussed on the basis of the nature of catalysts.

Synthesis of *N*-arylimidazoles using copper and its oxide

The synthesis of *N*-arylimidazoles can be achieved using copper or its oxide as a catalyst with or without ligand.

2-(Hydrazinecarbonyl)pyridine-*N*-oxides, proline, 4,7-dimethoxy-1,10-phenanthroline, 1,10-phenanthroline, and β -ketoimine ligands are used with copper powder or copper oxide to synthesise arylated imidazoles (**31**) in very good yields under ambient conditions for both electron-withdrawing and donating groups. For example, *N*-arylation of imidazoles (**30**) has been carried out with aryl halides (**29**) in water without an inert atmosphere in the presence of tetra-*n*-butyl ammonium bromide (TBAB) and the ligand 2-(2-hydrazinecarbonyl)pyridine-*N*-oxides^[56] [Scheme 8].

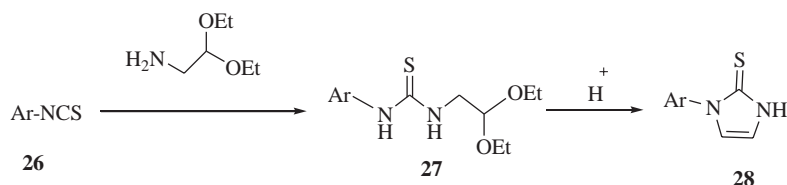
In an another approach, the ligand, 4,7-dimethoxy-1,10-phenanthroline has been found efficient for polyethyleneglycol (PEG) accelerated *N*-arylation (**34**) of variety of hindered and functionalized imidazoles and benzimidazoles (**32**) with aryl iodides and bromides^[57] (**33**) [Scheme 9].

Xue *et al.* have reported that β -ketoimine works as an efficient, inexpensive, and facile ligand for the Cu₂O-catalysed *N*-arylation (**37**) of imidazoles (**36**) with the electron rich, neutral, and electron deficient aryl bromides and iodides^[58] (**35**) [Scheme 10].

Cu₂O along with 1-(2-methylhydrazine-1-carbonyl)isoquinoline-2-oxide in aqueous medium proves to be a simple and efficient catalytic system for *N*-arylation (**40**) of imidazoles (**39**). Moreover, use of NaOH under above conditions brings about *N*-arylation of imidazole with 4-methoxyiodobenzene^[59] (**38**) [Scheme 11].

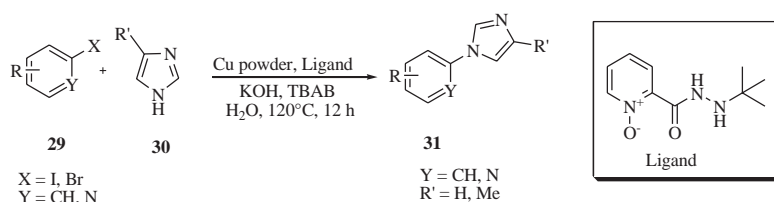
Various solvents such as water, dimethyl formamide (DMF), methanol, and ethanenitrile along with mild bases such as K₂CO₃, LiOH, and Cs₂CO₃ have led ligand free arylation to obtain desired products in considerably good yields.

One-pot synthesis of *N*-arylimidazoles (**43**) from indole, pyrazole, and imidazole (**42**) and aryl halides (**41**) has been carried out in water under air without using arylboronic acids, palladium, and base in the presence of copper based ionic liquid promoter^[60] [Scheme 12], similarly synthesis of

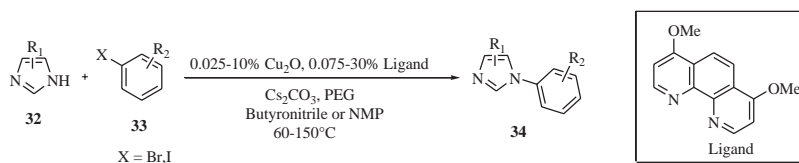


Ar: 2,4,6-Cl₃C₆H₂; 2,4,- C₆H₃(NO₂)(OMe);
3,4,5-(MeO)₃C₆H₂; 3,5-(CF₃)₂C₆H₃

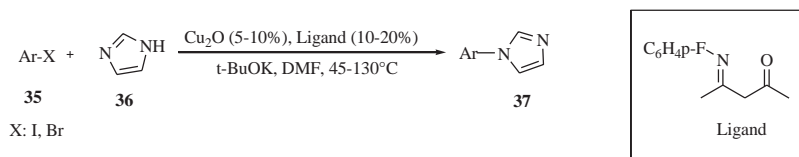
Scheme 7: *N*-Arylimidazole from aminoacetaldehyde diethyl acetal and aryl isothiocyanates



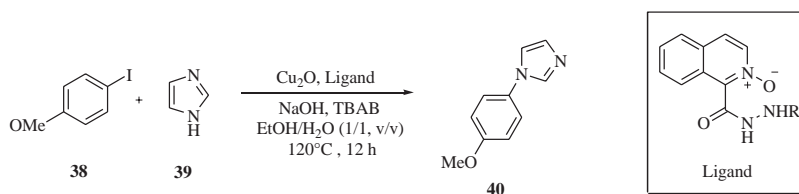
Scheme 8: *N*-Arylation of imidazoles using aryl halides



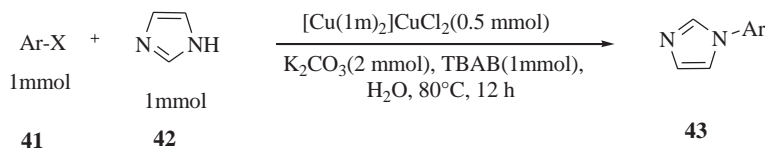
Scheme 9: *N*-Arylation of imidazoles using imidazoles, benzimidazoles, aryl iodides, and bromides



Scheme 10: *N*-Arylation of imidazoles using aryl bromides/iodides



Scheme 11: *N*-Arylation of imidazoles using 4-methoxyiodobenzene



Ar-X: *p*-I-C₆H₄NO₂, *p*-I-C₆H₄OCH₃, *o*-I-C₆H₄OCH₃, C₆H₅Br

Scheme 12: *N*-Arylation of imidazoles using aryl halides in ionic liquid

N-arylimidazoles (**43**) has been carried out with activated copper using LiOH as base from aryl halides (**44**) and imidazoles (**45**)^[61] [Scheme 13].

Wang *et al.*, in an Ullmann type reaction, established a protocol for Cu-catalysed *N*-arylation (**49**) of *N*-containing

heterocycles (**48**) with aryl or heteroaryl iodides and bromides (**47**).^[62] The procedure has shown a broad range of functional groups acceptance (electron donating as well as withdrawing) on both of the cross-coupling partners results in good to excellent yields [Scheme 14].

Various arylboronic acids (**50**) and amines have been converted to the corresponding *N*-arylazoles (**52**) and *N*-arylamines in a reaction efficiently catalysed with heterogeneous copper(I) oxide in methanol at room temperature under base free conditions^[63] [Scheme 15].

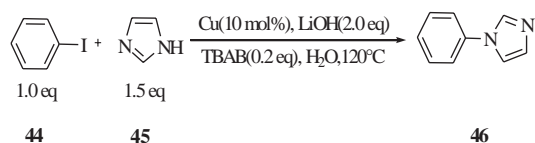
N-Arylimidazoles (**55-57**) can be obtained by treating imidazole (**54**) with 2,4-difluoroiodobenzene (**53**) in the presence of copper catalyst.^[64] The presence of both fluoro and iodo substituents has promoted S_NAr reaction at the C-F bond and copper-catalysed Ullmann-type coupling reaction at the C-I bond [Scheme 16].

Imidazoles (**58**) have been successfully arylated (**60**) with a wide range of chloro and fluoroarenes (**59**) using recyclable copper-exchanged fluorapatite catalyst in DMF solvent and K_2CO_3 as a base^[65] [Scheme 17].

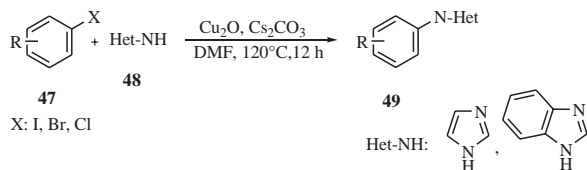
N-Arylated imidazoles (**63**) were synthesised from imidazole (**62**) and arylboronic acids (**61**) in the presence of a catalytic amount of SiO_2-NHC (*N*-Heterocyclic carbenes)-CuI in methanol at room temperature^[66] [Scheme 18].

Synthesis of *N*-arylimidazoles using copper halides

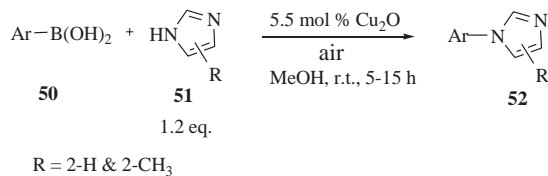
1-(5,6,7,8-Tetrahydroquinolin-8-yl)ethanone, 3-(diphenylphosphino)propanoic acid, oxazolidin-2-one,



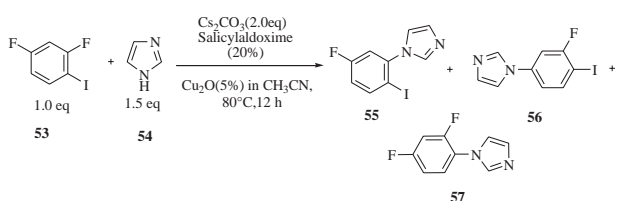
Scheme 13: *N*-Arylation of imidazoles using aryl iodide



Scheme 14: *N*-Arylation of imidazoles using benzimidazoles and aryl halides



Scheme 15: *N*-Arylation of imidazoles using arylboronic acids and amines



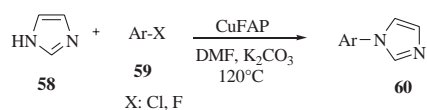
Scheme 16: *N*-Arylation of imidazoles using 2,4-difluoroiodobenzene

benzotriazole, (1*S*, 2*S*)-1,2-dimethylcyclohexane, and 1,10-phenanthroline, (1*R*, 2*R*)-cyclohexane-1,2-diamine, *N*(1),*N*(2)-dimethylethane-1,2-diamine, and quinolin-8-ol along with a base have been proven as highly efficient ligands affording *N*-arylated imidazoles under mild reaction conditions.

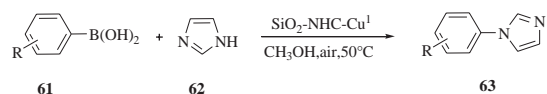
Ligand 1-(5,6,7,8-tetrahydroquinolin-8-yl)ethanone mediated copper catalysed approach for *N*-arylation (**66**) of azoles (**65**) with a variety of substituted aromatic bromides and iodides (**64**) under mild conditions has been employed with great functional group compatibility and excellent selectivity^[67] [Scheme 19].

N-Arylimidazoles (**69**) have been prepared under mild conditions using copper bromide as a catalyst from various *N*, *O*, and *S* nucleophilic reagents (**68**) with aryl halides (**67**) along through ligand 2-oxocyclohexanecarboxylate^[68] [Scheme 20].

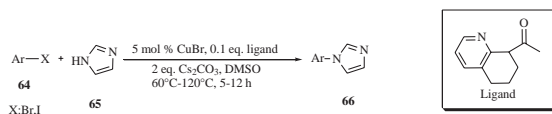
3-(Diphenylphosphino)propanoic acid is reported to be an efficient ligand for the copper-catalysed C–N coupling reactions for *N*-Arylation (**72**) of imidazole (**71**) with variety of substituted aryl iodides^[69] (**70**) [Scheme 21].



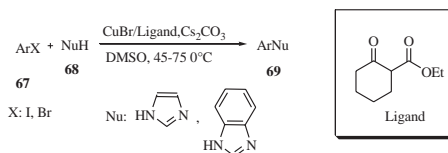
Scheme 17: *N*-Arylation of imidazoles using chloro/fluoroarenes



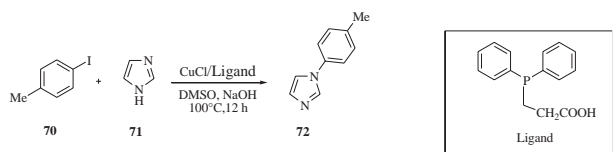
Scheme 18: *N*-Arylation of imidazoles using arylboronic acid



Scheme 19: *N*-Arylation of imidazoles using aromatic bromides/iodides with ligand 1-(5,6,7,8-tetrahydroquinolin-8-yl)ethanone



Scheme 20: *N*-Arylation of imidazoles using aromatic bromides/iodides with ligand 2-oxocyclohexanecarboxylate

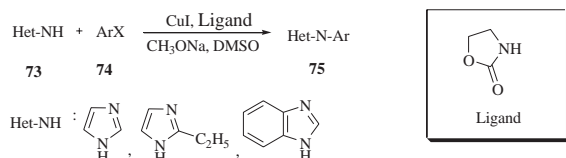


Scheme 21: *N*-Arylation of imidazoles using aryl iodides with ligand 3-(Diphenylphosphino)propanoic acid

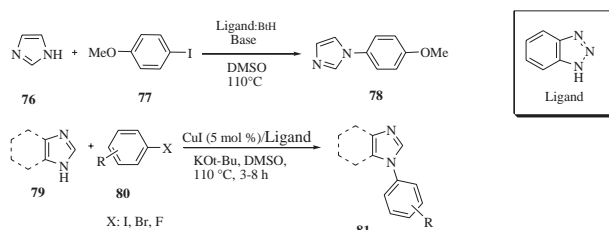
Oxazolidin-2-one acts as a facile ligand for N-arylation of pyrrole, indole, and imidazole (**73**) with aryl halides (**74**) in the Cu-I catalyzed reactions^[70] (**75**) [Scheme 22].

Another Cu-catalyzed *N*-arylation (**78**) of imidazoles (**76**) with substituted aryl iodide (**77**), a combination of copper iodide and benzotriazole (BtH) has proved to be an efficient and inexpensive catalyst^[71] [Scheme 23].

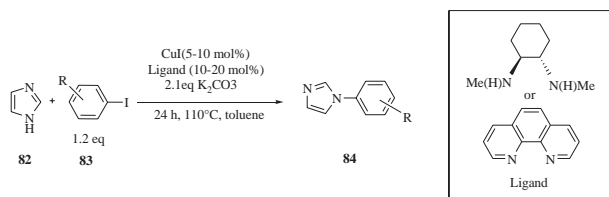
Synthesis of meta-bis(1-imidazolyl)benzenes (**84**) has been achieved through copper-catalysed coupling of variety of substituted aryl halides (**83**) and imidazole (**82**) in the presence of racemic trans-1,2-cyclohexanediamine^[72] [Scheme 24].



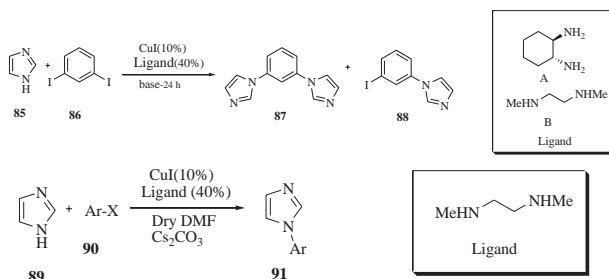
Scheme 22: *N*-Arylation of imidazoles using aryl halides with Ligand oxazolidin-2-one



Scheme 23: *N*-Arylation of imidazoles using *p*-methoxyiodobenzene



Scheme 24: *N*-Arylation of imidazoles using 1,3-diiodobenzene



Scheme 25: *N*-Arylation of imidazoles using 1,3-diiodobenzene

Alcalde *et al.* have chosen the ligand combination of racemic *trans*-1,2-cyclohexanediamine and *N,N*-dimethylenediamine for coupling of imidazole (**85**) and 1,3-diiodobenzene (**86**) for an Ullmann-type condensation that affords bis(imidazole) (**87**, **88**), the simple structural motif based on meta-bis(1-imidazolyl) benzenes in excellent yields^[73] [Scheme 25]. Furthermore, the first ever synthesis of sterically hindered 1-mesitylimidazole has been reported with 50% isolated yields after chromatographic purification.

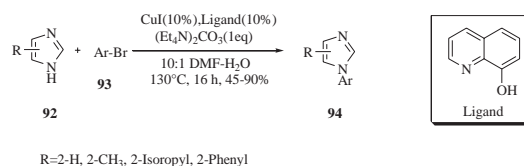
In a heterogeneous system consisting of tetraethylammonium carbonate as a base, supported by 8-hydroxyquinilone as ligand and water as cosolvent, CuI has successfully catalysed *N*-arylation (**94**) of imidazoles (**92**) with aryl bromides^[74] (**93**) [Scheme 26].

Hexamethylenetetramine when used as a ligand, transformed imidazole (**96**) when treated with aryl chlorides and bromides (**95**) into their corresponding *N*-arylated imidazole^[75] (**97**) [Scheme 27].

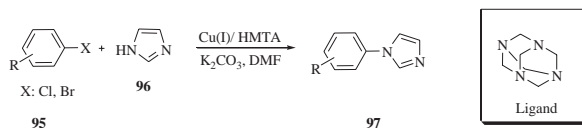
The ratio of imidazole-phenylboronic acid has played a crucial role for coupling of imidazole (**99**) with arylboronic acid **98** in protic solvent^[52] in the ratio of more than 1.2:1 in developing *N*-arylimidazole (**100**) [Scheme 28].

High chemoselectivity has resulted into *N*-arylation (**103**) of imidazole (**102**) with a variety of aromatic bromides and iodides^[76] (**101**) [Scheme 29].

An efficient, facile, mild, and base free *N*-arylation (**106**) of imidazoles (**105**) with arylboronic acids (**104**) using copper-exchanged fluorapatite catalyst have been preceded at room temperature^[77] [Scheme 30].

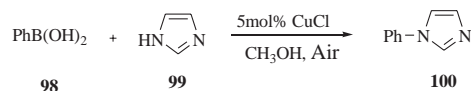


Scheme 26: *N*-Arylation of imidazoles using aryl bromide

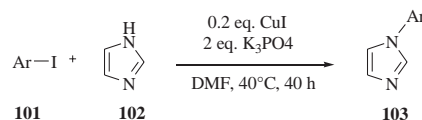


R=4-NH₂, 4-OCH₃, 4-COCH₃, 4-CHO, 2-OCH₃, 1-Phenylethanol, 4-OH, 4-Br, 4-Cl, 2 & 4-NO₂-C₆H₄.

Scheme 27: *N*-Arylation of imidazoles using hexamethylenetetramine



Scheme 28: *N*-Arylation through coupling of imidazole with arylboronic acid



Scheme 29: *N*-Arylation of imidazoles using aryl bromides/iodides

Synthesis of *N*-arylimidazoles using copper acetate

Copper acetate along with ligands 4,4'-dimethyl-2,2'-bipyridine and sparteine in water as solvent have been used to achieve arylation of imidazoles under mild conditions. In one of the greener copper catalyzed and efficient methods, *N*-arylimidazoles (**109**) have been obtained by oxidative coupling of arylboronic acids (**107**) with imidazoles (**108**) in water and base-free conditions using an amphiphilic surfactant^[78] [Scheme 31].

Mao *et al.* have developed arylated imidazoles (**112**) from aryl iodide (**110**) and imidazole (**111**) with catalyst $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$ / (-) sparteine^[79] [Scheme 32].

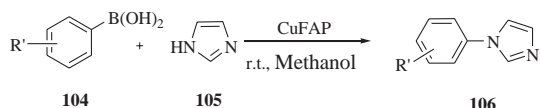
Copper acetate has catalysed *N*-arylation (**115**) of imidazole (**114**) in the absence of ligands as well as through aryl/heteroaryl C-N cross-coupling approach through the arylboronic acid (**113**)/cupric acetates but it requires longer reaction time^[47] [Scheme 33].

Unsymmetric arylimidazolium and triazolium salts (**118**) can be synthesised from *N*-substituted imidazoles (**116**) and diaryliodonium salts (**117**) using a copper catalyst^[80] [Scheme 34].

Miscellaneous Synthesis of *N*-arylimidazoles

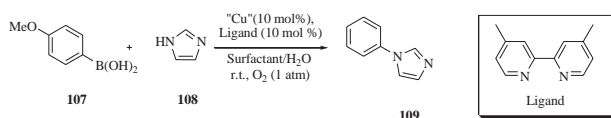
N-Arylation can also be carried out through cobalt-catalysed coupling of diaryliodonium salts with imidazole, microwave-assisted methods (with/without solvent), use of ionic liquids and nano particles, copper-complex catalysed, and photocatalysed reactions. Most of these methods are environment friendly, highly efficient and inexpensive which furnished in good yields.

N-Arylation (**121**) of imidazole (**120**) with diaryliodonium salt (**119**) was carried using Cobalt(II) acetate as a catalyst and K_2CO_3 as a base in DMF^[81] [Scheme 35].

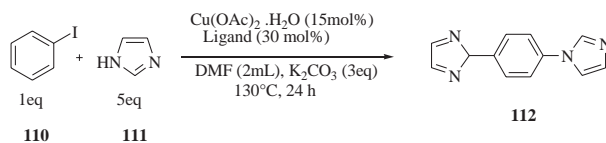


R': H, CH₃, OCH₃, NO₂, F, Cl, CF₃

Scheme 30: *N*-Arylation of imidazoles using copper-exchanged fluorapatite



Scheme 31: *N*-Arylation of imidazoles using arylboronic acids and amphiphilic surfactant



Scheme 32: *N*-Arylation of imidazoles using aryl iodide and catalyst

Solvent-free coupling reaction between imidazole (**122**) and differently substituted aryl bromides (**123**) has been carried out under microwave irradiation using L-amino acids and copper salts as catalysts^[82] to get *N*-arylimidazole (**124**) [Scheme 36].

Yang *et al.* have reported a ligand-free copper-catalysed Ullmann reaction in tetraethyl orthosilicate and microwave-assisted synthesis of *N*-arylimidazoles^[83] (**127**) from 1*H*-imidazoles (**126**) and aryl halides (**125**) [Scheme 37].

The use of ionic liquid $[\text{Bmim}]_4\text{BF}_4$ as a reaction medium has led to the reduction in reaction time with increased yields of *N*-arylimidazoles^[84] (**130**) from *N*-heteroaryls (**129**) and aryl/heteroaryl halides (**128**) [Scheme 38].

Collman *et al.* have synthesized and applied a series of binuclear bis- μ -hydroxo copper(II) complexes to the coupling of imidazoles (**131**) with arylboronic acids (**132**) in dry CH_2Cl_2 at room temperature under oxygen to get *N*-arylation of imidazole (**133**).^[50] In a similar approach, the reaction is reported to be selectively performed in an *N*-methyl-2-pyrrolidone and water (NMP/ H_2O) mixture at ambient temperature and atmosphere using $[\text{Cu}(\text{OH})\text{TMEDA}]_2\text{Cl}_2$ dimer as the catalyst without the addition of base^[53] [Scheme 39].

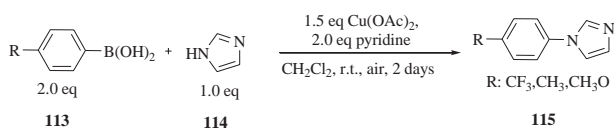
A novel, green photocatalytic route for *N*-Arylation (**139**) of imidazoles (**138**) with arylboronic acids (**137**) involving the C-N cross-coupling reactions has been reported over a Cu/graphene catalyst^[85] [Scheme 40].

Synthesis of *N*-arylimidazoles (**142**) from aryl halides (**140**) and imidazole (**141**) under mild conditions using copper nanoparticles (CuNPs) has been developed in excellent yields^[86] [Scheme 41].

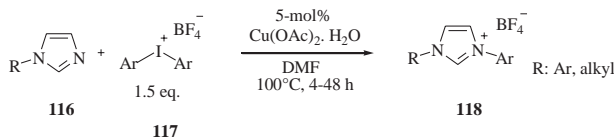
Several bases have been tested for the Cu-grafted nanomaterial Cu-MPTA-1 (mesoporous poly-triallylamine) *N*-arylation (**145**) of imidazoles (**143**) with iodobenzene (**144**) in water as a solvent. The best results have been obtained with iodobenzene using KOH in water^[87] at 120°C for 12 h [Scheme 42].

Magnetically recoverable nanocatalyst-copper nanoparticles on nanosized silica-coated maghemite (CuNPs/Mag Silica) have been used for the copper-catalysed *N*-arylation (**148**) of imidazoles (**147**) with aryl bromides and iodides (**146**) under ligand-free conditions^[88] [Scheme 43].

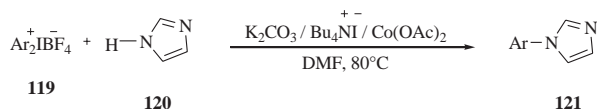
An environmentally sound process for *N*-arylation (**151**) of imidazoles (**150**) with aryl halides (**149**) using copper oxide/acetylene black (CuO/AB) in toluene and a base has afforded the product with 100% yield^[89] [Scheme 44].



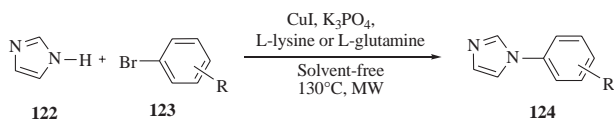
Scheme 33: N-Arylation of imidazoles using arylboronic acid/cupric acetate



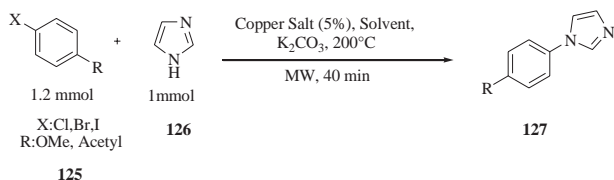
Scheme 34: N-Arylation of imidazoles using diaryliodonium salts and copper acetate



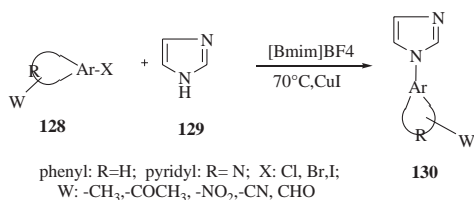
Scheme 35: N-Arylation of imidazoles using diaryliodonium salt and cobalt acetate



Scheme 36: N-Arylation of imidazoles using aryl bromides under solvent free condition



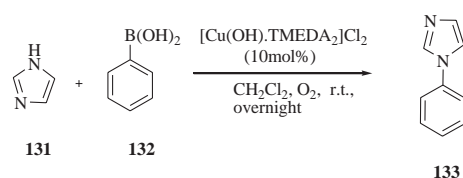
Scheme 37: N-Arylation of imidazoles through Ullmann reaction



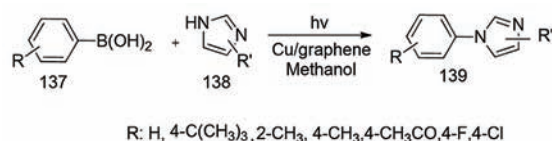
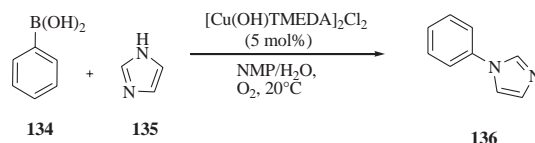
Scheme 38: N-Arylation of imidazoles using ionic liquid

Liu *et al.* have reported the arylation (154) of benzimidazole (153) with a variety of aryl iodides (152), bromides 152 promoted by salen copper (II) complex in moderate to excellent yields under mild conditions^[90] [Scheme 45].

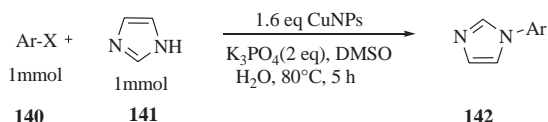
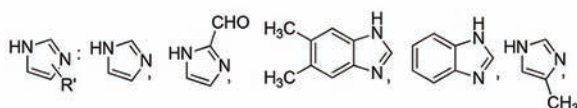
A novel strategy adopted by Wang *et al.*, involving mild reaction conditions involving aryl halides (155) and N-heteroarylimidazole (156) catalyzed by a sulfonato-Cu



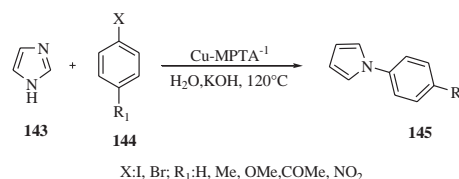
Scheme 39: N-Arylation of imidazoles using arylboronic acids



Scheme 40: A greener approach towards N-arylation of imidazoles using arylboronic acids



Scheme 41: N-Arylation of imidazoles using copper nanoparticles



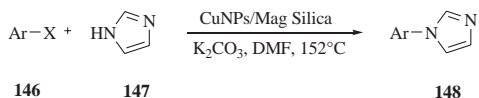
Scheme 42: N-Arylation of imidazoles using Cu-grafted nanomaterial

(salen) complex to get N-arylimidazole^[91] (157) [Scheme 46].

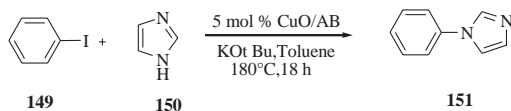
Copper-complexes (mononuclear [CuLCl]ClO₄ and binuclear [Cu₂L₂(μ-1,1-N₃)₂](ClO₄)₂) catalysed N-arylation (160) of imidazole (158) with aryl iodides (159) (electron-poor or electron-rich) resulted in a highly efficient, user-friendly, sustainable, and low-cost protocol^[92] [Scheme 47].

Boron trifluoride derivative of N-arylated imidazole N-oxides (164) was obtained by condensing aromatic amines (161) with formaldehyde (162) and butane-2,3-dione monooxime (163) in the presence of boron trifluoride etherate in acetic acid or chloroform. These could be easily reduced to their corresponding N-arylated imidazoles (165) by reducing with iron in acetic acid^[93] [Scheme 48].

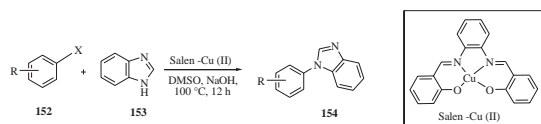
N-Substituted 2,4-diarylimidazoles (**170**) were synthesized under solvent-free conditions in a one-pot



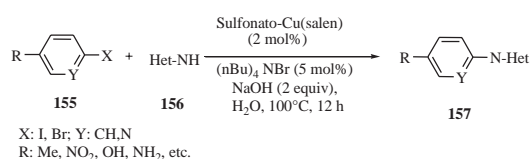
Scheme 43: *N*-Arylation of imidazoles using CuNPs on nanosized Mag Silica



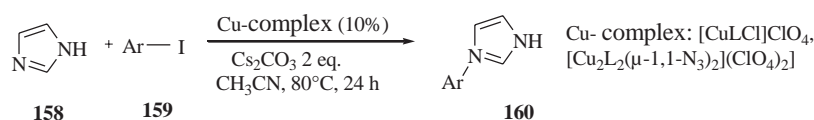
Scheme 44: *N*-Arylation of imidazoles using CuO hollow nanospheres on acetylene black



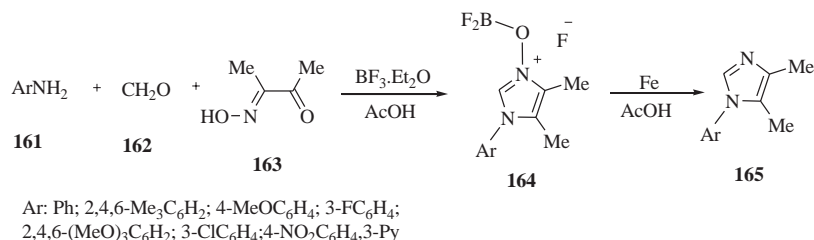
Scheme 45: *N*-Arylation of imidazoles using salen copper (II) complex



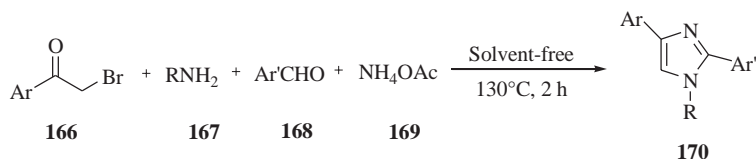
Scheme 46: *N*-Arylation of imidazoles using sulfonato-Cu salen complex



Scheme 47: *N*-Arylation of imidazoles using mononuclear and binuclear Cu-complexes



Scheme 48: Synthesis of *N*-arylated imidazole using metallic iron



Scheme 49: Synthesis of *N*-arylated imidazole by multicomponent reaction

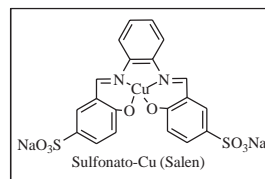
reaction using four components: 2-bromoacetophenone (**166**), an aldehyde (**168**), a primary amine (**167**), and ammonium acetate^[94] (**169**) [Scheme 49].

CONCLUSION

Imidazole and its derivative *N*-substituted imidazole attracted various researchers to design and synthesize such compounds due to their growing medicinal value. The present review has covered recently reported important approaches toward the synthesis of *N*-substituted imidazole involving two basic approaches, i.e., constructing imidazole moiety from open-chain substrates and *N*-arylation of imidazoles. Since variously substituted *N*-arylimidazoles are associated with diverse biological activities, this review might stimulate further studies for the development of potential compounds as a replica of therapeutic agents.

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CONFLICTS OF INTEREST

The authors declare that they have read policy and guidelines of the journal and there are no conflicts of interest.

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